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## FERROELASTIC AF3 AND A1- xA'xF3 FLUORIDES WITH A VF3-RELATED STRUCTURE

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Some trifluorides AF \_\_ (A = A1,Sc,Ti,V,Cr,Fe,Co,Ga,In) crystallize at room temperature with a rhombohedral VF \_\_ 3 structure derived from the ReO \_ cubic-type.

<sup>3</sup> By direct synthesis in sealed gold tubes or by fluorinating mixtures of AF and  $Tl_2O_3$ , various solid solutions A A' F have been characterized (the A' and A' cations have been chosen in connection with the progressive increase of their size) :

 $\begin{array}{l} \operatorname{Al}_{1-\mathbf{x}} \operatorname{Cr}_{\mathbf{x}} \operatorname{F}_{3}, \operatorname{Cr}_{1-\mathbf{x}} \operatorname{Ti}_{\mathbf{x}} \operatorname{F}_{3}, \operatorname{Cr}_{1-\mathbf{x}} \operatorname{In}_{\mathbf{x}} \operatorname{F}_{3}, \operatorname{Ti}_{1-\mathbf{x}} \operatorname{In}_{\mathbf{x}} \operatorname{F}_{3} : 0 \leqslant \mathbf{x} \leqslant 1 \\ \operatorname{Al}_{1-\mathbf{x}} \operatorname{Ti}_{\mathbf{x}} \operatorname{F}_{3} : 0 \leqslant \mathbf{x} \leqslant 0.82 \quad \text{and} \operatorname{In}_{1-\mathbf{x}} \operatorname{Tl}_{\mathbf{x}} \operatorname{F}_{3} : 0 \leqslant \mathbf{x} \leqslant 0.85 \end{array}$ 

They crystallize also with a rhombohedral symmetry. A phase transition has been detected for each composition by micro-DTA, microcalorimetry, optical measurements and thermal evolution study of the X-ray spectrum. The high-temperature form is cubic (Fm3m).

The ferroelastic domain structure and its disappearance above the transition temperature allowed us to predict this transition to be of ferroelastic type. The room temperature spontaneous strain e and the Curie temperature T (610 K < T < 1224K) have been correlated to the size of the various cations. The values of e (300K) and T increase as r at decreases in a same family, for example for the 3d-transition metal fluorides AF (A = Ti,V,Cr,Fe). The smallest A<sup>3</sup> cations allow a larger rotation of the fluorine atoms around the three-fold axis of the octahedron which is responsible for the strain of the crystal network.

When two different cations  $A^{3+}$  and  $A^{*}$  fill the cationic six-fold coordinated sites (for the composition  $A_{0,5}A^{*}_{0,5}F_{3}$ ) the variation of  $e_{s}$  and T has also been discussed with regards to the size of  $A^{*}$  and  $A^{*3+}$  cations.